

The Structure of Kreysiginine (Alkaloid CC-21)

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THREE of the major alkaloids from *Kreysigia multiflora* Reichb. (Liliaceae) have been proved¹ to be homoaporphines and so to be members of the growing new class² of 1-phenethylisoquinoline alkaloids. The spectroscopic data for kreysiginine, a further alkaloid from this plant,³ clearly showed it to be of a different type. Moreover, kreysiginine has now been found to be structurally identical but

enantiomeric with alkaloid CC-21 from *Colchicum cornigerum*^{4†} (Sweinf.) Täckh. et Drar. (Liliaceae). Our results establish structure (I) for (+)-krey-siginine which is a further 1-phenethylisoquinoline derivative and related, as a ring B homologue, to the morphine group of alkaloids [cf. thebaine(II)].

Mass spectrometry supported the molecular formula³ C₂₁H₂₇NO₅ for kreysiginine (M⁺ 373),

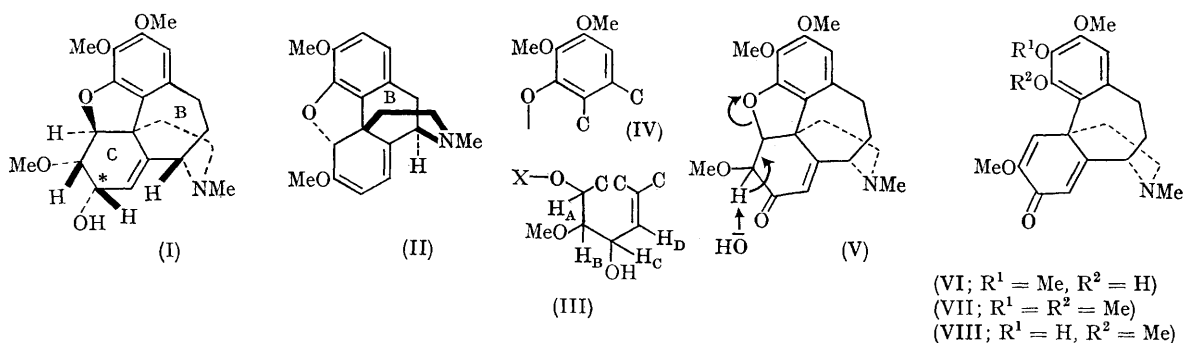
<i>N.m.r. spectra; τ values (CDCl₃)</i>							
Substance	N-Me	O-Me	-CH _B OMe	-CH _C OR	-CH _A O-	CH _D =C	Aromatic
Kreysiginine (I)	7.45	6.51, 6.24, 6.13	6.81, 6.78, 6.27, 6.68 <i>J</i> _{AB} 9.5, <i>J</i> _{BC} 4.0	5.81, 5.77, 5.75, 5.71 <i>J</i> _{BC} 4.0, <i>J</i> _{CD} 6.0	5.45, 5.36 <i>J</i> _{AB} 9.5	4.37, 4.31 <i>J</i> _{CD} 6.0	3.86
<div style="display: flex; justify-content: space-around; align-items: center;"> <div style="text-align: center;">↔</div> <div style="text-align: center;">↔</div> </div>							
Couplings demonstrated for above case							
<i>O</i> -Acetylkreysiginine	7.42	6.61, 6.23, 6.11	Obscured by OMe 6.71, 6.68 <i>J</i> _{BC} 4.0	4.59, 4.55, 4.53, 4.49 <i>J</i> _{BC} 4.0, <i>J</i> _{CD} 6.0	5.48, 5.39 <i>J</i> _{AB} 9.5	4.37, 4.31 <i>J</i> _{CD} 6.0	3.85

n.m.r. confirmed the presence of one *N*-methyl and three *O*-methyl groups (Table), and i.r. proved the absence of carbonyl groups. A set of coupled signals was present in the 100 Mc./sec. spectrum which established the sequence (III); double-resonance experiments demonstrated all the spin-spin couplings indicated in the Table. Acetylation of kreysiginine proved its nitrogen to be tertiary and the n.m.r. spectrum of the resultant *O*-acetylkreysiginine ($M^+ 415$) confirmed sequence (III) by showing the expected downfield shift of the signal assigned to $>CHOAc$. Kreysiginine shows u.v. absorption corresponding to a 1,2,3-trialkoxybenzene chromophore and it possesses only one aromatic proton; partial structure (IV), or an isomer, is thus indicated. The foregoing functional groups and the molecular formula require kreysiginine to be pentacyclic and one of these rings is assigned to a cyclic ether system involving the oxygen X-O-CH of (III) in accordance with the inert nature of this residue.

When partial structures (III) and (IV) are considered biogenetically, constitution (I) becomes a probable one for kreysiginine; proof of this structure was obtained as follows. Mild Jones

oxidation of kreysiginine afforded dehydro-kreysiginine (V) showing enone absorption (ν_{\max} 1690 cm^{-1}). Treatment of the enone with base opened the oxide bridge yielding a dienone (VI, $M^+ 371.1733$; $C_{21}H_{25}NO_5$, 371.1724), ν_{\max} 1663, 1638, 1613 cm^{-1} , which was *O*-methylated with trimethylanilinium hydroxide to generate the ether (VII, $M^+ 385.1888$; $C_{22}H_{27}NO_5$, 385.1889). This product, $[\alpha]_D -169^\circ$ (CHCl_3), was identical by full spectroscopic and chromatographic comparison with *O*-methylandrocybine,² $[\alpha]_D -295^\circ$ (CHCl_3) of rigorously established structure and absolute configuration (VII). Structure (I) thus represents (+)-kreysiginine with the relative configurations still to be considered. The difference in optical rotation for the product (VII) prepared from androcybine² (VIII) and from kreysiginine is a reflection of the partial racemic nature of the latter whose $[\alpha]_D$ has varied over the range $+29$ to $+89^\circ$ for different batches. Similarly, alkaloid CC-21† is a partial racemate with $[\alpha]_D -42^\circ$ (CHCl_3) for the only specimen available; this alkaloid thus contains the enantiomer of structure (I) in admixture with some (+)-form.

The large coupling (9.5 c./sec.) between H_A and



† Isolation and elucidation of the structures of many other alkaloids from *C. cornigerum* will be described in a joint paper from the Olomouc and Liverpool groups.

H_B of kreysiginine (see III) requires a *trans*-diaxial relationship for these protons which is in agreement with the illustrated configurations on a half-chair ring c. The hydroxy-group must then be set axial to account for J_{BC} 4 c./sec. (H_B and H_C in axial-equatorial relationship) and this agrees with J_{CD} 6.0 c./sec., the dihedral angle H_C-H_D being *ca.* 30°. Structure (I) is now a complete representation for (+)-kreysiginine.

Retention of oxygen at the starred carbon of ring c (I) for a molecule having an oxide bridge is in contrast to the situation obtaining in the morphine group of alkaloids.⁵ The biosynthetic implications of this observation are being explored by tracer experiments with *K. multiflora*.

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¹ A. R. Battersby, R. B. Bradbury, R. B. Herbert, M. H. G. Munro, and R. Ramage, *Chem. Comm.*, 1967, 450.

² A. R. Battersby, R. B. Herbert, L. Pijewska, and F. Šantavý, *Chem. Comm.*, 1965, 228; A. R. Battersby, R. B. Herbert, and F. Šantavý, *Chem. Comm.*, 1965, 415; A. R. Battersby, R. B. Herbert, E. McDonald, R. Ramage, and J. H. Clements, *Chem. Comm.*, 1966, 603; A. C. Barker, A. R. Battersby, E. McDonald, R. Ramage, and J. H. Clements, *Chem. Comm.*, 1967, 390.

³ G. M. Badger and R. B. Bradbury, *J. Chem. Comm.*, 1960, 445.

⁴ M. Saleh, S. El-Gangihi, A. El-Hamidi, and F. Šantavý, *Coll. Czech. Chem. Comm.*, 1963, 28, 3413; see also H. Potěšilová, J. Hrbek jun., and F. Šantavý, *ibid.*, 1967, 32, 141.

⁵ A. R. Battersby, Tilden Lecture, *Proc. Chem. Soc.*, 1963, 189; D. H. R. Barton, Hugo Muller Lecture, *Proc. Chem. Soc.*, 1963, 293; D. H. R. Barton, G. W. Kirby, W. Steglich, G. M. Thomas, A. R. Battersby, T. A. Dobson, and H. Ramuz, *J. Chem. Soc.*, 1965, 2423.